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AP Chemistry Chemical Equations Worksheet Write the balanced chemical equation (excluding spectator ions) underneath each reaction description, and answer the question. (a) A solution of ammonia is added to a dilute solution of acetic acid. Identify the conjugate acid-base pairs in this reaction. (b) Solutions of sodium hydroxide and acetic acid are mixed. Acetic acid is a weak acid. If ...

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6. A 50.0L reaction vessel contains 1.00mol N<sub>2</sub>, 3.00mol H<sub>2</sub>, and 0.500mol NH<sub>3</sub>. Will more ammonia, NH<sub>3</sub>, be formed or will it dissociate when the mixture goes to equilibrium at 400.0C?

### AP Chemistry Equilibrium Worksheet

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